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Investigation of Intumescent Fabric Coatings For Protection Against **Thermal Radiation and Flame**

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ABSTRACT

Evaluation of intumescent and miscellaneous organic fabric coatings for protection of personnel indicates that varying amounts of energy attenuation are provided and, therefore, some degree of protection is possible.

A new area of exploration has been opened with the current investigation of the use of intumescent materials on clothing as a possible means of protecting personnel against flame and thermal radiation. The National Aeronautics and Space Administration recently held a meeting at Moffet Field, Calif. (1), where some new intumescent paints containing aromatic nitroamine bisulfates were described. These materials were pointed out to be very effective in protecting various inanimate substrates, including canvas, against jet fuel fires and simulated solar radiation. More recently Parker et al. discussed the properties of p-nitroaniline bisulfate (2). However, no information is available on the application of these intumescent materials to clothing for protection of personnel against thermal radiation and flame. Similarly, very little information is available in the literature on any other intumescent materials applied to clothing. A few foam-forming materials were evaluated in 1960 by A. D. Little, Inc. (3), as additives to 5- and 9-oz cotton poplin. These additives, consisting generally of a polymer and blowing agent or a pre-foamed polymer, did not appreciably change the heat transfer properties of the poplin.

The idea of using intumescent materials on clothing as a means of protecting personnel against flame and thermal radiation is attractive for several reasons: (1) A large amount of protective char is generated by some intumescent formulations during exposure to heat; (2) the protec-

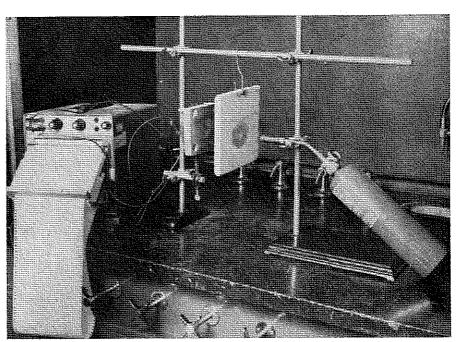


Figure 1 Flame Test Apparatus

tive char is developed only when needed; (3) the protection capability is present at all times; and (4) the rate of heat transfer through the coatings and char appears relatively low.

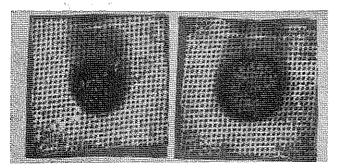
However, there are some potential disadvantages to intumescent coatings on fabrics when these materials are considered for personnel protection: (1) The coatings can appreciably increase the weight of a fabric substrate and thereby the load carried by a person; (2) the air permeability of fabrics can be reduced appreciably (except when coatings are applied to mesh materials), thereby reducing the comfort of the individual; and (3) the coatings can require several seconds of heating to form large volumes of char during which

R70-70

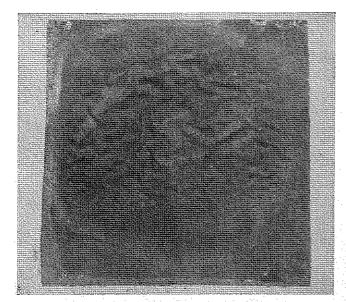
TABLE I Coatings on Non-Fabric Materials

Coating Thickness (mils)*

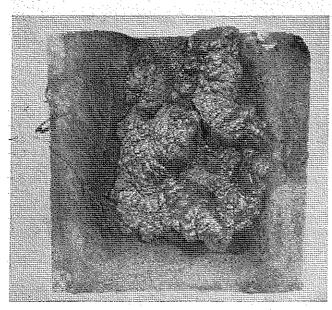
		Brush Method	Poured Method	
Formula	Material		Average	Range
I (o-nitroaniline in	Aluminum sheet	5	70	40-100
nitrocellulose)	Asbestos board	5	70	40-100
II (o-nitroaniline in polyurethane)	Aluminum sheet	15-20	60	50-70
	Asbestos board	15-20	60	50-70



Nylon mesh with Coating II, over uncoated fiber N-70. Carbon arc exposure: left, 20 cal; right, 10.3 cal

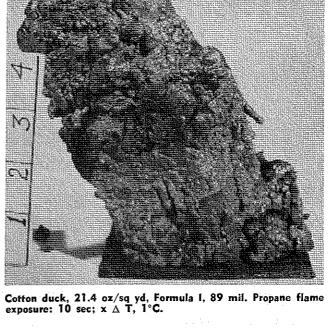


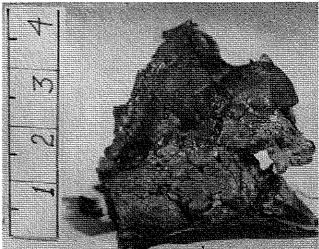
Unexposed Formula I on cotton duck



time the temperature behind the sample may rise high enough to cause a burn.

In spite of these potential disadvantages it seemed worthwhile to evaluate selected coatings and fabric combinations, especially since to our knowledge intumescent coatings had never been evaluated on heat resistant fabrics made with Nomex* nylon fiber and the experimental Fiber N-





Cotton/rayon cloth, Formula I, 26 mil. Propane flame exposure: 5 sec; Δ T, 28°C.

Figure 2
Intumescence Exhibited by Selected Materials
Exposed to Carbon Arc and Propane Flame

Cotton/rayon cloth, Formula I, 231 percent add-on. Propane flame exposure: 5 sec; Δ T, 63°C.

70. Some non-fabric substrates were also used for comparison. The evaluations were carried out with the aid of both a simple flame test apparatus designed and assembled in this laboratory and the U. S. Army Natick Laboratories (NLABS) arc-image furnace.

*E. I. du Pont de Nemours & Co., Inc., trade name for a fire-resistant nylon fiber.

EXPERIMENTAL

The heat-resistant fabrics and other materials selected for evaluation were:

- Wool: blanket, 13.8 oz/sq yd, 35 percent reprocessed; and serge 9.6 oz/sq yd.
- 2. Fiber N-70: poplin, 4.2 oz/sq yd.
- 3. Nomex: herringbone twill, 3.3 oz/sq yd.

- 4. Nylon 66: 9 oz/sq yd mesh, Type MIL-C-8061.
- 5. Beta-glass cloth, factory-coated with carboxy nitroso rubber: cloth, 27% oz/sq yd; coating, 21/4 oz/sq yd of cloth.
- 6. Aluminum: 5 mil, polished sheet.
- 7. Asbestos: ½ in. asbestos board, unpainted.
- 8. Nitroso rubber: mesh 48 oz/sq yd and 102 oz/sq yd.
- 9. Cloth, cotton warp, rayon fill, 7.6 oz/sq yd.
- 10. Cotton duck, 21.4 oz/sq yd.
- 11. Cotton duck with FWWMR treatment, base fabric 9.85 oz/sq yd, coated weight 13.9 oz/sq yd.

All of these materials were Military Olive Green (OG) in color except for the coated β -glass, which was white. All but the nitroso rubber mesh materials and the experimental fiber N-70 fabric were available commercially. The nitroso rubber for the mesh was compounded with commercial fillers, such as oxides of iron, chromium and zinc, to achieve colors approaching that of the Military OG. The rubber was cured according to conventional curing processes in the form of 46- and 80-mil tensile sheets. These sheets were perforated with a cork borer (size number one) at the rate of 16 holes per sq in. to obtain the mesh.

The potentially intumescent coatings consisted of o-nitroaniline, sulfuric acid, and various binders, namely, nitrocellulose, polyurethane rubber, polyvinyl alcohol, and zein. In addition, char and/or smoke-forming coatings of the protein zein and a corn dextrin were prepared. Since o-nitroaniline in any form is especially toxic to humans according to Sax (4), this compound and the coatings containing it were handled with gloves in a well-ventilated area.

To make the coatings the following procedures were undertaken:

Coating I — o-Nitroaniline (27.5g) was added to a mechanically stirred solution consisting of nitrocellulose (20.7g commercial nitrocellulose cement, specifically, Ambroid Liquid Cement, 29 percent solids, Ambroid Co., Inc., Weymouth, Mass.) and methyl ethyl ketone (MEK) (19g, 23.8 ml) until all particles had dissolved. Sulfuric acid. (19.7g, 11.5 ml) was dissolved in 13.1g (16.4 ml) absolute ethanol and then added dropwise to the nitrocellulose solution as the stirring continued. During the addition, the bisulfate salt of o-nitroaniline precipitated and heat was liberated. Vigorous stirring continued for two hours, with occasional pauses to determine if the salt would remain in suspension. Since it did not MATERIAL: Fiber N-70

Coatings and/or Overlayers Indicated on Graph

THERMAL SOURCE: Carbon Arc

EXPOSURE TIME: 1 Second

IRRADIANCE LEVEL:

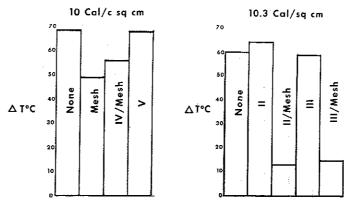


Figure 3 Relative Temperature Rise (ΔT) Behind Coated Materials Exposed to Thermal Radiation

MATERIAL: Fiber N-70

Coatings and/or Overlayers Indicated on Graph

EXPOSURE TIME: 1.1 Seconds THERMAL SOURCE: Carbon Arc

THERMAL SOURCE: Carbon Arc IRRADIANCE LEVEL: 10 Cal/sq cm **EXPOSURE TIME: 1 Second** IRRADIANCE LEVEL: 10 Cal/sq cm Nitroso Rubbe IVA/IV ∆ T°C

∆T°C

Nitroso Rubber Wooi Blanket Wool Serge Cloth Nitroso Rubber/B-Glass <u>=</u> Ē 8

MATERIALS: Miscellaneous

with Coatings and/or

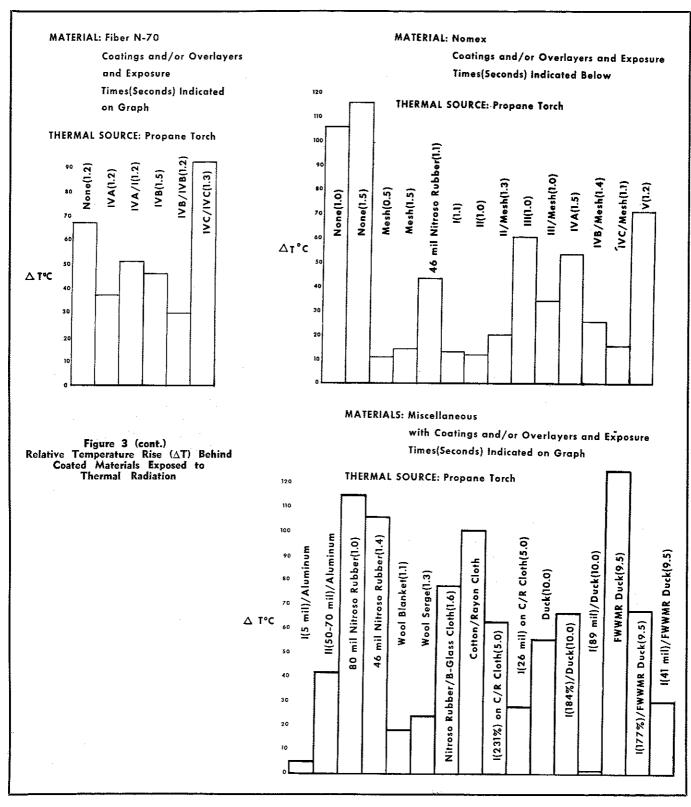
Indicated on Graph

Overlayers

the solution was constantly stirred during the coating process.

Coating II — Polyurethane rubber cement, 46.3g (Bostic® adhesive, 15 percent solids, USM Corp., Cambridge, Mass.), was diluted with 6.5g (7.3 ml) of tetrahydrofuran (THF). To this solution was added 27.5g of o-nitroaniline with mechanical stirring until all particles dissolved. This was followed by dropwise addition of 19.7g (11.5 ml) of concentrated sulfuric acid with continued stirring. As in Coating I, the bisulfate salt precipitated and constant stirring was needed to keep it in suspension.

Coating III - o-Nitroaniline, 27.5g, was mixed by mechanical stirring with 287.1g of a polyvinyl alcohol solution (Partal Film No. 10, 7 percent Solids, Boatex Fiberglas Co., Inc., Needham, Mass.) after which 19.7g (11.5 ml) of concentrated sulfuric acid was added with continued



stirring. During each of these steps only part of the o-nitroaniline dissolved. It was completely dissolved by heating the mixture to 89°C on a steam bath with constant stirring. The solution was then cooled, whereupon a very fine precipitate appeared. This precipitate remained suspended for more than 24 hours. No additional solvent was used.

Coating IVA — Zein, 10g, (Corn Products Sales Co.) was dissolved

with occasional swirling in a 70 percent by volume isopropyl alcohol solution. The alcohol solution was prepared by mixing 70cc (54.6g) of the alcohol with 30 ml (30g) of water. When the zein was completely dissolved, 2.74 ml more water was added. The resulting solution was slightly viscous.

Coating IVB — This solution was prepared similarly to coating formula IVA except that 1.37 ml of 38 percent

formaldehyde solution was substituted for the extra water. The weight of formaldehyde was 0.5g, that is, 5 percent based on the weight of zein.

Coating IVC — A zein solution was prepared according to coating formula IVA to which was added 9.2g of onitroaniline. The mixture was heated under hot tap water to help dissolve the o-nitroaniline. It was then cooled and 3.8 ml of concentrated sulfuric acid was added dropwise with stir-

ring. All solids dissolved completely. Coating V — Dextrin, 20g, (Hercules, Inc.) was added in portions with stirring to 80 ml of water until all particles were dissolved.

Test Sample Preparation

The fabric and non-fabric materials were prepared for testing by cutting them into approximately 4 x 4 in. specimens and coating them directly with the mixtures and solutions described without a "sealer" or finish coat. The coatings were applied to the materials with a paint brush except in a few instances where they were poured on the materials. These coating methods were adequate for preliminary evaluations but they did not produce coatings of uniform thickness, partly because there were relatively large particles of solid suspended in many of the coating mixtures. The safety precautions used in handling the o-nitroaniline-containing coatings were also used in handling the respective coated materials.

The coated test samples are described in Tables I and II. Uncoated control samples of each material were also prepared. The wool blanket, wool serge, and nitroso rubber were tested in an uncoated state only.

The NLABS arc-image furnace, which was used for the irradiation screening tests, was described previously by McCue (5). The flame test apparatus constructed in this laboratory consisted of a sample holder, propane torch, shutter, and recording potentiometer. These components were assembled as indicated in Figure 1.

TABLE II
Coatings on Fabric Materials

	•			
Coating	Material	Method of Drying	Number of Coats	Add-on by weight (%)
I (o-nitroaniline in nitrocellulose)	Nylon mesh	air	1	119.0
II (o-nitroaniline in polyurethane)	Nomex Fiber N-70 Nylon mesh* Nylon mesh**	air "	1 1 1 1	239.0 128.0 121.0 112.0
III (o-nitroaniline in polyvinyl alcohol)	Nomex Fiber N-70 Nylon mesh* Nylon mesh**	air ,, ,,	1 1 1	126.0 74.0 48.5 45.0
IVA (zein)	Fiber N-70 Fiber N-70 Fiber N-70 Nylon mesh	air oven oven oven	1 1 2 1	13.6 10.2 25.2 8.1
IVB (zein with formaldehyde)	Fiber N-70 Fiber N-70 Fiber N-70	air oven oven	1 1 2	15.0 14.3 32.0
IVC (zein with o-nitroaniline)	Fiber N-70 Fiber N-70 Nylon mesh	air "	1 2 1	26.5 48.3 20.7
V (dextrin)	Fiber N-70 Nomex Nylon mesh	air "	1 1 1	44.2 77.0 24.8
Nitroso rubber	8-glass	_	_	7.5

^{*}Used with untreated Nomex **Used with untreated Fiber N-70

The sample holder consisted of a transite block containing a copperconstantan thermocouple embedded in its front surface. The holder and the propane torch were positioned so that the tip of the torch was 5 in. from the front surface of the holder. The shutter, suspended between the holder and the torch, was a piece of

 $\frac{1}{2}$ -in. asbestos board. The torch was a Turner LP-5 propane torch with an LP-603 pencil point burner. When exposing a sample, the temperature rise (Δ T) behind the sample was registered in millivolts on a Honeywell Electronik 19 recorder which was connected to the thermocouple in the sample holder.

Temperature

TABLE III

Behavior of Materials Irradiated in the Carbon Arc Image Furnace

Material	Coating formula	Add on (%)	Flame during exposure	Smoke formation	Appearance after exposure	rise behind material (ΔT °C)
Nylon Mesh/ Fiber N-70	no coating	no add-on	No	None	mesh fibers fused; some char at high ΔT	23.7 to 54.8
Wool Blanket	no coating	no add-on	No	much white smoke	some char; back side intact	28.3
Nitroso rubber mesh, 48 oz.	no coating	no add-on	No	much smoke	surface etched and covered with friable black particles	16.9 to 19.5
eta-glass	carboxy- nitroso rubber (CNR)	7.5	No	very little	no change	54.8
Nylon Mesh/ Fiber N-70	Formula I/ no coating	119/0	Yes	yellow smoke	some intumesced char on mesh	28.3
Nylon Mesh/ Fiber N-70	Formula II/ no coating	112/0	Yes	yellow smoke	some intumesced char on mesh	13.2
Nylon Mesh/ Fiber N-70	Formula III/ no coating	45/0	Yes	yellow smoke	mesh charred/ Fiber N-70 slightly charred	16.7

Test Procedures

Temperature rise (ΔT) determined in the NLABS are-image furnace. A sample was attached with two rubber bands to the sample holder so that there was as much contact as possible between the sample and the thermocouple in the holder. Each sample, consisting of either a single cloth swatch or a mesh swatch over a cloth swatch, was exposed at approximately 10 cal/cm² for 1 sec. The

appearance of each sample before and after exposure and the behavior of each sample during exposure were observed. Also, the temperature rise at the back surface of the sample was recorded in millivolts. During the exposures, dark glasses were worn by personnel to protect their eyes from the radiation while observing the samples.

Temperature rise (ΔT) determined in a propose torch flame. A sample was attached as described above to

the sample holder and the recorder was adjusted to the appropriate speed and attenuation settings. The torch was then ignited and the valve opened wide. Each sample was then exposed to the flame on a reverse countdown of three. At 3, all was in readiness; at 2, the recorder was turned on; and at 1, the shutter was removed and a stop watch was started simultaneously. At zero, the shutter was closed and the watch was stopped. As in the above procedure, the appear-

TABLE IV

Behavior of Materials Exposed to a Propane Torch Flame

Material	Coating formula	Add-on	Exposure time (sec)	Combustion during exposure	Smoke formation	Appearance after exposure	Temperature rise behind material (ΔT °C)
Nomex	no coating	no add-on	1-1.5	yes, almost none at lowest ΔT	none	large hole; some melting; damage small at lowest ΔT	18 to 115.5
Nylon mesh/ Nomex	no coating	no add-on	0.5-1.5	none	none	mesh melted; holes in Nomex at higher ΔT	10.5 to 35, possibly greater
Wool blanket	no coating	no add-on	1.1	yes	none	nap burned off; some char; back surface un- changed	13
Wool serge	no coating	no add-on	1.3	none	none	some char; back surface unchanged	24
Aluminum	I	5 mil coating	1-7	none	yellow smoke	intumesced char ¾" to ¼" high	0-5*
Asbestos	Ï	about 5 mil coating	10	none	none except when flame stopped	intumesced char 1/8" high	not recorded
Aluminum	I	70 mil coating	. 15	none	yellow smoke	intumesced char 5-6" high	not recorded
Asbestos	I i	50-100 mil coating	17	none	a little; some yel- low and white smoke	intumesced char 1½" high	not recorded
Nylon mesh/ Nomex	Formula I/ no coating	119%/0%	1-5	none	yellow smoke	some intumesced char at low ΔT , many mesh holes plugged; holes in Nomex at high ΔT	11.5 to 14.0
Asbestos		about 5 mil coating	10	none -	puff of yellow smoke when flame stopped	intumesced char	not recorded
Aluminum	11	50-70 mil uneven coating	30+	none	yellow smoke	intumesced char 9/16" high; hole in aluminum	42*
Asbestos	11	50-70 mil coating	24	none	puff of yellow smoke when flame stopped	intumesced char 3/4" high	not recorded
Nomex	II	239%	1	none	yellow smoke	some intumesced char about 1/4" high	10.5 to 13
Nylon mesh/ Nomex	Formula II/ no coating	121%	1-1.5	sometimes	yellow smoke	intumesced char; at higher ΔT some mesh holes filled with char and some holes in Nomex	14 to 26.5
Nylon Mesh/ Nomex	Formula III no coating	/ 48.5%	1	none	yellow smoke	char and melting; holes in Nomex	29- 37.5

^{*}Thermocouple not in good contact with aluminum sheet.

ance of a sample before and after exposure and its behavior during exposure were noted. Also, the temperature rise as indicated by the thermocouple was determined. Most of the exposures lasted 1-2 sec, but a few samples were exposed up to 30 sec. This procedure was rapid and provided reasonably consistent exposure times for easy comparison of results.

RESULTS AND DISCUSSION

Intumescence was observed in varying amounts only on the materials having Coatings I and II containing the nitrocellulose and polyurethane binders, respectively. Examples of this intumescence are shown in Figure 2. The other materials produced merely smoke or vapors and/or unfoamed char. These types of behavior occurred in both the irradiance and the flame tests. Since Goodwin et al. (1) reported intumescent action in nitroamine bisulfates alone, the above evidence indicates that some binders can interfere with this phenomenon as recognized by Parker et al (2). Thus the binder could play an important role in the intumescent reaction.

The amount of intumescence observed during the tests increased with the exposure, that is, with the irradiance level or the number of cal/cm²/sec in the irradiance test and

with the exposure time in the flame test. This increase was noticed, for example, with the coated nylon mesh which, it was hoped, would show enough intumescence to fill the mesh holes. However, most of the char foamed perpendicular to the plane of the samples toward the energy source rather than laterally in the plane of the sample, and thus most of the holes did not fill in. The greatest amount of intumescence was observed with a very thick coating of nitrocellulose formulation on aluminum sheet. When exposed to the flame for about 15 sec this sample increased in thickness from 70 mils to 5 or 6 in., an 8,000 percent increase. This result agreed well with the results of Goodwin et al. (1) and served as a standard for comparison.

Several of the materials screened in the irradiance and flame tests showed a high degree of energy attenuation as indicated by the relatively low temperature rise (AT) behind the samples during exposure. These "low ΔT " materials are shown in Tables III and IV with their ΔT values and other observations of their behavior. The remaining materials exhibited ΔT values from 30 up to 150°C. Examples are shown in Figure 3. It is interesting to note that several of the low ΔT materials exhibited no intumescence and some were uncoated.

A satisfactory ΔT value with re-

spect to personnel protection in these screening tests was arbitrarily selected as a maximum of 10 to 20°C, the lower temperature being preferable. Although this rise in temperature is insignificant for inanimate objects, it may or may not produce a burn on human skin. On this basis, the best materials in the irradiation test were nitroso rubber mesh by itself and the composites of nylon mesh with a polyurethane or polyvinyl alcohol coating over Fiber N-70. The best materials in the flame test were the wool blanket, Nomex, nylon mesh over Nomex, Nomex with the polyurethane coating, and the composites of nylon mesh with a nitrocellulose or polyurethane or zein-onitroaniline coating over Nomex. However, these materials did not necessarily exhibit ΔT values consistently in the 10 to 20°C temperature range. Thus, only a limited amount of protection for personnel would be obtainable with these materials, but their properties are attractive for applications where personnel are not concerned. Table V summarizes the behavior and potential protection for personnel afforded by all the combinations of coatings and fabrics tested.

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TABLE V

Behavior and Protection Afforded by Coated Fabric Combinations and Miscellaneous Materials

Materials	Exposure source	Behavior and protection
1. Nitroso rubber mesh	carbon arc	
Nitroso rubber mesh over Nomex	flame	smoke formation, limited protection no protection
3. Nitroso rubber coated glass cloth	carbon arc and flame	smoke sometimes observed, no protection
4. Nomex alone and nylon mesh over Nomex	flame	limited protection
5. Nomex with poly- urethane (very high add on)	flame	smoke formation, limited protection
6. Nylon mesh with nitrocellulose, poly- urethane or zein-nitro- aniline coatings over Nomex	flame	char and smoke, limited protection
7. Nylon mesh with poly- urethane and polyvinyl alcohol coatings over Fiber N-70	carbon arc	char and smoke, limited protection
8. Wool blanket	carbon arc and flame	smoke and char, no or limited protection,
9. Wool serge	carbon are and flame	respectively smoke and char, no
10. Dextrin on Fiber N-70 and Nomex	carbon arc and flame	protection char, no protection
11. Zein or zein-formal- dehyde on Fiber N-70 and Nomex	carbon arc and flame	gases, no protection

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